Ionization from Solar Pumped Metastable Levels of Atomic Samarium

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Work Sponsored at NRL by 6.1 Base Program



AFRL MOSC Experiment (Radar Data from ALTAIR)





Ionization Processes in Samarium Vapor

- Why did MOSC Samarium Not Produce Predicted Density Levels?
- Samarium Atom Photo Chemistry (NRL CRM)
 - Sm Energy Levels
 - Sm Metastable Level Pumping in Sunlight (Important)
 - Samarium Photo-Ionization (Important but Slow)
 - Samarium Associative Ionization with Atomic Oxygen
 - Reaction Energy
 - Weakly Exothermic from Ground State Sm (7F) Metastable Levels
 - Strongly Exothermic from Higher Sm(9H, 7H, ...) Metastable States
 - SmO⁺ Production (Autoionization) and Loss (Recombination-Important)
 - Samarium Reaction with Diatomic Oxygen (Important)
- 3-D Time Dependent Predictions for MOSC Release
- Data Acquired During AFRL MOSC Experiment for Comparison
 - Initial Electron Production Inventory from NRL CERTO Beacon
 - Altair Radar Map of Electron Density
 - AFRL Spectrogram of Optical Emissions
- Conclusions

Conceptual Samarium Photo Chemistry



Solar Resonance Fluorescence +hvSolar Photo-Ionization hvAuto-Ionization with Atomic Oxygen 0Dissociative Recombination e^{-} Oxidation with Molecular Oxygen -0_2



All Known Samarium Atomic Levels



Normalized Equilibrium Populations of the Metastable Levels of Samarium with Direct Solar Illumination with Auto-Ionization Dependence on Energy



Samarium Neutral Diffusion Based on the MSIS Atmosphere for 9 May 2013

$$D_{Sm} = \left(\sum_{j \neq Sm} 1/D_{Smj}\right)^{-1} \text{ where } D_{Smj} = \frac{3}{32 f r_{Smj}^2 n_j} \left(1 + \frac{m_{Sm}}{m_j}\right)^{1/2} \left(\frac{8kT_{Sm}}{\pi m_{Sm}}\right)^{1/2} \text{ and } j = 0, N_2, \text{ or } O_2$$



 $D_{Sm} = 5.83 \times 10^8 \text{ cm}^2/\text{s}$ at 171 km Altitude

(Latest) Time Dependent Computation of Sm⁺ and SmO⁺ Ions for Sm Release in Sunlight and Autoionization Reaction with O

• Solar Photoionization Reaction

 $Sm + hv_{Sun} \xrightarrow{\beta_{SmSun}} Sm^+ + e^- \text{ rate: } \beta_{SmSun} = 0.00442 \ s^{-1}, \tau_{SmSun} = k_{SmSun}^{-1} = 220 \text{ s}$

• Metastable State Autoionization Reaction for Release at 171 km Altitude

$$Sm^{*} + O \xrightarrow{k_{Sm+O}} SmO^{+} + e^{-} + \Delta E_{\alpha} \text{ rate: } k_{Sm+O} = \phi_{SmExo} \overline{\sigma_{Sm+O}} v$$

$$\phi_{SmExo} = 0.104, \ \sigma_{Sm+O} \square \ 5 \times 10^{-15} \, cm^{2}, v \square \sqrt{kT_{O} / m_{O}} \square \ 718 \text{ m/s for } T_{O} = 1000K$$

$$k_{Sm+O} \square \ 3.73 \times 10^{-10} \text{ cm}^{3}/\text{s}, n_{O} = 6.8 \times 10^{9} \text{ cm}^{-3}, \beta_{Sm+O} = k_{Sm+O} n_{O} = 2.54 \text{ s}^{-1}$$

- Samarium Oxidation* $Sm[\alpha] + O_2 \xrightarrow{k_{Sm+O_2}} SmO + O$ $k_{Sm+O_2} \square 5.1 \times 10^{-10} \text{ cm}^3/\text{s}, n_{O_2} = 6.4 \times 10^8 \text{ cm}^{-3}, \beta_{Sm+O_2} = k_{Sm+O_2}n_{O_2} = 0.32 \text{ s}^{-1}$
- Dissociative Recombination Reaction $SmO^{+}[{}^{6}\Gamma] + e^{-} \xrightarrow{k_{SmO^{+}e^{-}}} Sm[{}^{7}F] + O[{}^{3}P]$ rate: $k_{SmO^{+}e^{-}} \approx 10^{-7} \text{ cm}^{3}s^{-1}$

*Note: Sm + O_2 Reaction from M. L. Campbell, Temperature-Dependent Rate Constants for the Reactions of Gas-Phase Lanthanides with O_2 , J. Phys. Chem. A, 1999, 103 (36), pp 7274–7279

3D Numerical Model of Sm Release Photochemistry

- Neutral and Ion Equations with Chemical Reactions
 - Neutral Samarium, Samarium Monoxide Ion, Samarium Ion, Samarium Monoxide, Electrons

$$\begin{aligned} \frac{\partial N_{sm}}{\partial t} &= \frac{D_1}{R} \frac{\partial}{\partial R} \left(R \frac{\partial N_{sm}}{\partial R} \right) + D_1 \frac{\partial^2 N_{sm}}{\partial z^2} - \beta_{sm} N_{sm} + k_{smO^+e^-} N_{smO^+} N_e \\ \frac{\partial N_{smO^+}}{\partial t} &= D_1 \frac{\partial^2 N_{smO^+}}{\partial z^2} + \beta_{sm+O} N_{sm} - k_{smO^+e^-} N_{smO^+} N_e, \beta_{sm+O} = k_{sm+O} n_O \\ \frac{\partial N_{sm^+}}{\partial t} &= D_1 \frac{\partial^2 N_{sm^+}}{\partial z^2} + \beta_{smSun} N_{sm} \\ \frac{\partial N_{smO}}{\partial t} &= \frac{D_1}{R} \frac{\partial}{\partial R} \left(R \frac{\partial N_{smO}}{\partial R} \right) + D_1 \frac{\partial^2 N_{smO}}{\partial z^2} + \beta_{sm+O_2} N_{sm}, \beta_{sm+O_2} = k_{sm+O_2} n_{O_2} \\ N_e &= N_{sm^+} + N_{smO^+}, \ \beta_{sm} \equiv \beta_{smSun} + \beta_{sm+O} + \beta_{sm+O_2} \end{aligned}$$

- Cylindrical Coordinates with **z** along **B**
- Numerical Solution by Expanding Boundary Coordinate Transform



Central Cloud Density for Samarium Release Including *Recombination* Sm Release Product Neutrals and lons



New 3-D Model for Samarium Release at t = 20 s



3-D Model for Samarium Release at t = 100 s



Resonance Fluorescence of Samarium Atoms and Atomic Ions



Estimated Total Electron Content Yield for MOSC Samarium Releases

MOSC CERTO Beacon to Rongelap, 1 and 9 May 2013



AFRL MOSC Experiment, ALTAIR – Launch 2



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Central Cloud Density for Samarium Release Including *Recombination* Sm Release Product Neutrals and lons



MOSC Optical Spectra (From Todd Pedersen and Jeff Holmes, AFRL)



Summary on MOSC Samarium Release

- Factors that Control the Ionization from Sm Release in Sunlight
 - Formation Metastable States
 - Photo-ionization
 - Atomic Oxygen Reaction
 - Recombination of Samarium Monoxide Ion (Depletes Electrons)
 - Reaction of Samarium with Diatomic Oxygen (Depletes Samarium)
- Physics Based Modeling of MOSC Sm Release
 - Predictions of Metastable Level Population, (Sm and Sm⁺) Optical Spectra
 - Time Dependent Predictions of Ion Compositions and Electron Density
 - Spectral Lines for Sm, Sm+ and SmO
 - Future Work (Model Validation and Prediction for Future Experiments)
 - Compare with Beacon, Radar and Optical Observations
 - Compare with AFRL Empirical Model
 - Compute HF Refraction Off Model Electron Clouds
- Conclusions
 - CRM Model Is Converging on Accurate Solutions
 - MOSC Used Critical Diagnostics
 - Visible Spectrograph Yields Neutral and Ion Composition
 - Incoherent Scatter Radar Yields Long Term Electron Production
 - Radio Beacon Instrument Yields Initial Electron Production
 - SmO⁺ + e⁻ Recombination is Exothermic and Very Reactive
 - Sm + O Reaction is Slightly Exothermic and Very Reactive
 - Sm Release Probably Will Produce Few Ions Without Sunlight